

## "Formal"-Ru(II) Catalyzed Cycloadditions of 1,6-Diynes to Arenealkylidenes

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## "Formal"-Ru (II) catalyzed cycloadditions

Transition-metal-catalyzed cycloadittion reactions constitute powerful methods for the construction of complex polycyclic systems<sup>1</sup>

We recently described a new "formal" ruthenium-catalyzed [4+2+2] cycloaddition of 1,6-diyne 1 to *cis,cis*-1,4-disubstituted conjugated dienes 2 to give conjugated 1,3,5-cyclooctatrienes  $3.^2$ 



When arenealkylidenes were used as dienes in the "formal" ruthenium-catalyzed [4+2+2] cycloaddition, different cycloadducts were obtained according to the aromatic diene:

In the case of the styrene 4, aryltriene 5 was obtained, which was cycliced to arylciclohexadiene 6 upon heating.

However, when propenylfurane **7** was used, the cyclooctatriene **9** was obtained as a sole product. Most probably, compound **8** could be the initial product of the "formal" [4+2+2] cycloaddition, followed by a [1,5]-hydrogen shift to the final product **9**.







reductive liminatior 6πе disrotatory CI oxidative  $L = CH_3CN$ coupling Y = CH, n = 2 $X = C(CO_2Me)_2$ 8πе conrotator A C1 Mé [1,5]-H 9 1.2-insertion β- elimination Y=O, n=1Y=O, n=1Y=CH, n=2Y= O, n=1 Figure 1

The observed products and the position of the double bonds could be explained if the mechanism showed in the figure 1 is operating.

The energy surfaces theoretically calculated for these "formal" cycloadditions reactions agree with the experimental results showed above. (Energy in Kcal/mol)



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